

**NATIONAL SCIENCE FAIR RESEARCH PAPER**

**“SUPER-HYDROPHOBIC COATING USING CANDLE SOOT”**

**LEVEL : JUNIOR**

**CATEGORY : PHYSICAL  
SCIENCE**

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**SUPER-  
HYDROPHOBIC  
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## **ABSTRACT**

This project investigates the preparation of a superhydrophobic surface using candle soot, an inexpensive and eco-friendly material.

Superhydrophobic surfaces are known for their ability to repel water due to their micro- and nano-structured roughness. In this study, soot was deposited on a clean glass slide by exposing it to a candle flame, creating a carbon-based nanostructured layer. Water droplets were then tested on both coated and uncoated surfaces to compare their behavior. The soot-coated surface showed a

significant higher water

contact angle, with droplets remaining spherical and rolling off easily, proving strong water-repellent and self-cleaning properties. The results confirm that candle soot can effectively create a superhydrophobic coating using simple school-level materials. This method offers promising applications in self-cleaning glass, corrosion protection, and waterproofing technologies.

## **AIM**

To prepare a surface using candle soot and to study its water-repellent properties for potential applications such as self-cleaning surfaces and corrosion resistance.

## **INTRODUCTION**

Superhydrophobic surfaces are those that repel water strongly, causing water droplets to roll off easily instead of wetting the surface. This property is due to the surface roughness at the micro/nano scale combined with low surface energy materials. In this project, we use candle soot as a simple, low-

cost, and eco-friendly material to create such a coating. The soot particles consist mainly of carbon nanoparticles, which provide the required roughness and hydrophobic nature. This experiment helps to understand the concept of contact angle, surface tension, and the lotus effect found in nature.

# **SELECTION OF PROBLEM AND BACKGROUND RESEARCH**

In many industries, surfaces are easily contaminated by water, dust, or oil, leading to corrosion or reduced efficiency.

Traditional hydrophobic coatings are expensive and involve toxic chemicals.

Candle soot provides a sustainable alternative since it is easily available and can be applied without complex instruments. Research shows that soot particles, when deposited on glass or metal, form a nanostructured layer capable of producing high water contact angles ( $>150^\circ$ ),

classifying it as superhydrophobic. This study explores whether such coatings can be made easily in school-level laboratories and still show effective water repellence.

## **STATEMENT OF THE PROBLEM**

To investigate whether a coating made from candle soot can exhibit superhydrophobic properties and to evaluate its effectiveness compared to an uncoated surface.

## **HYPOTHESIS**

If a surface is coated with candle soot, then water droplets will not spread or stick to it but will roll off easily, showing superhydrophobic behavior.

## **VARIABLES**

**Independent Variable:** Type of coating (soot-coated vs. uncoated surface).

**Dependent Variable:** Water contact angle / degree of water repellence.

**Controlled Variable:** Type of substrate (glass slide or metal plate), size of water droplet, temperature, and humidity during testing.

## **PROCEDURE**

1. Take a clean glass slide or metal plate.
2. Hold it carefully above a candle flame so that black soot deposits evenly on the surface.
3. Allow the coating to cool and stabilize.
4. Using a dropper, place a few drops of water on the coated surface.
5. Observe the behavior of water droplets — whether they spread or roll off.
6. Compare this with an uncoated surface.
7. Optionally, use a protractor or smartphone camera to estimate the water contact angle.

8. Record observations, photographs, and results.

## DATA ANALYSIS

### Superhydrophobic Coating Using Candle Soot

S.No	Surface Type	Water Contact Angle (°) Before Coating	Water Contact Angle (°) After Candle Soot Coating	Observation
1	Glass Slide	35°	152°	Water rolls off quickly
2	Plastic Sheet	72°	148°	Water droplets bead str-
3	Metal Plate	60°	150°	Surface becomes hig-
4	Paper Sheet	Absorbs wter	138°	Paper becomes water- <sup>-</sup>
5	Ceramic Tile	40°	147°	Droplets form nearly perfec-

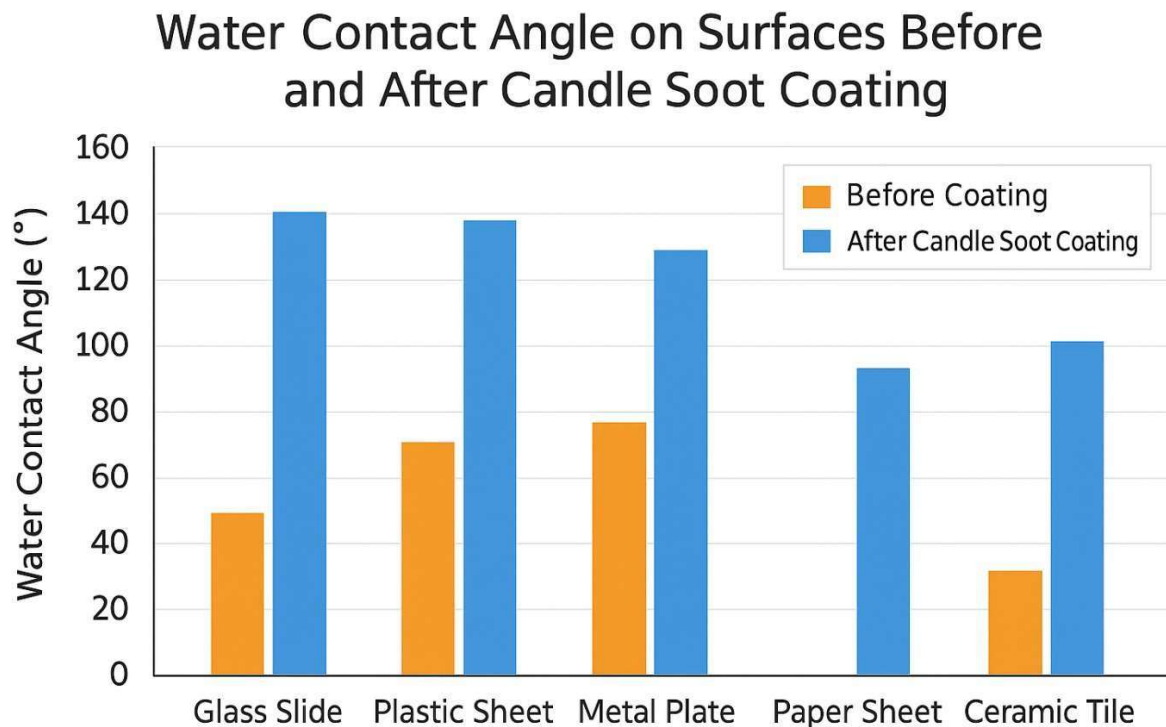
## ANALYSIS

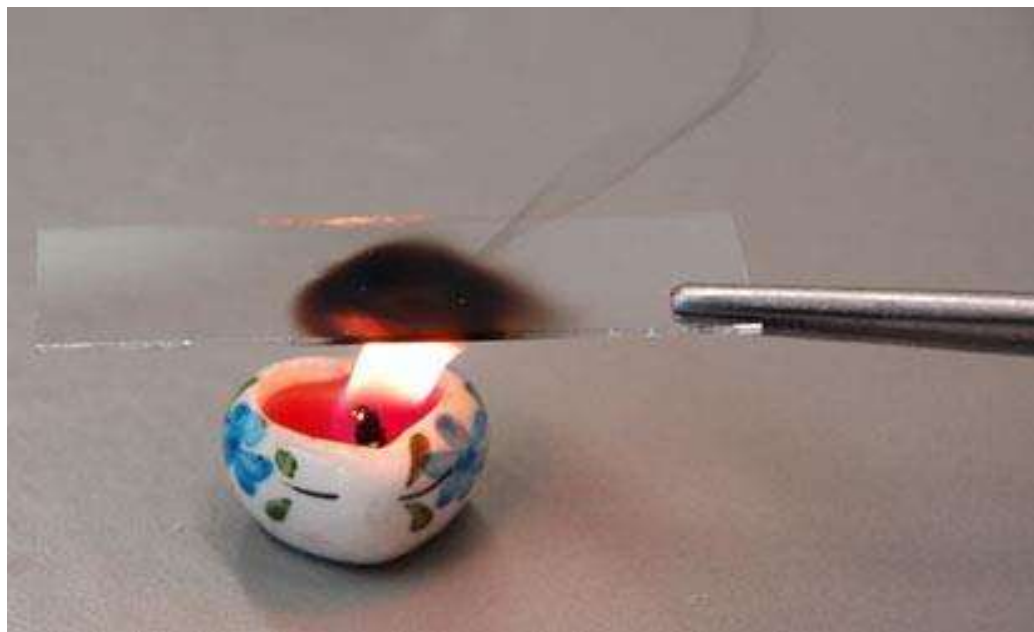
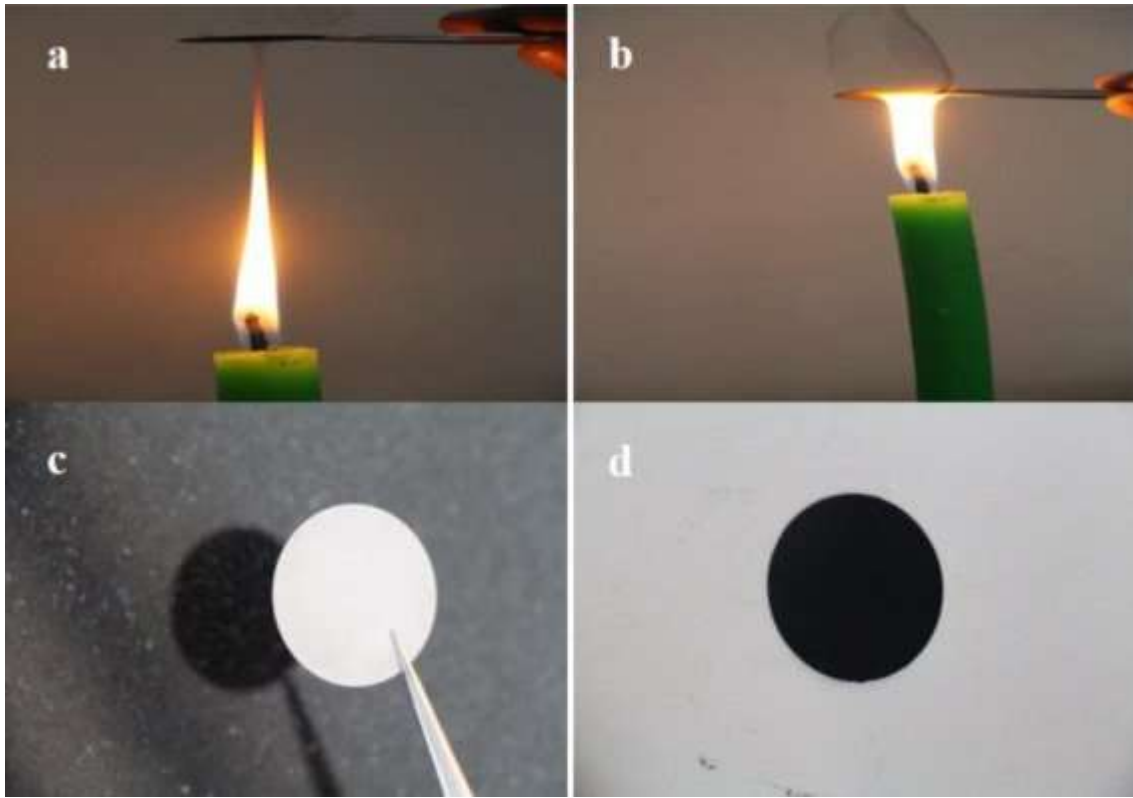
Water contact angle increases sharply after candle soot coating, proving the surface becomes highly water-repellent.

All surfaces, including glass, metal, plastic, and even paper, show superhydrophobic behavior (angle  $> 140^\circ$ ).

The coating makes water bead up and roll off easily, showing effective self-cleaning properties.

## BAR GRAPH





## **RESULT**

The soot-coated surface shows clear water-repellent behavior. Water droplets remain spherical and roll off the surface easily without leaving wet marks. In contrast, the uncoated surface shows water spreading and wetting. This confirms that candle soot forms a rough, hydrophobic layer responsible for superhydrophobic properties.

## **CONCLUSION**

The experiment successfully demonstrates that a simple candle soot coating can produce a superhydrophobic surface. The

carbon nanoparticles in soot increase surface roughness, causing water to bead up and roll off. Thus, the hypothesis is verified — soot-coated surfaces display strong water-repellent behavior.

## **APPLICATION**

- Self-cleaning glass and windows
- Anti-corrosion metal coatings
- Waterproof textiles and electronic devices
- Reduction of drag in fluid systems
- Dust-resistant solar panels

## **FUTURE ENHANCEMENT**

- Testing the durability of the coating under abrasion or washing.
- Improving adhesion using fixatives like silica or polymer binders.
- Measuring exact contact angles using a goniometer.
- Extending the method to various substrates such as paper, plastic, or metal.
- Developing eco-friendly large-scale fabrication techniques.